# **Effect of Formulation Factors on In Vitro Transcorneal Permeation of Gatifloxacin From Aqueous Drops**

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#### ABSTRACT

The purpose of this research was to optimize the formulation factors for maximum in vitro permeation of gatifloxacin from aqueous drops through excised goat cornea and to evaluate the permeation characteristics of drug from selected marketed eyedrop formulations. Permeation studies were conducted by putting 1 mL of formulation on the cornea  $(0.67 \text{ cm}^2)$  fixed between the donor and receptor compartments of an all-glass modified Franz diffusion cell and measuring gatifloxacin concentration in the receptor (containing normal saline under stirring) by spectrophotometry at 291.5 nm, after 120 minutes. Raising the drug concentration of the drops increased the drug permeation but decreased the percent permeation and the in vitro ocular availability. Raising the pH of the formulation from pH 5 to 7.2 increased both the drug permeation and the in vitro ocular availability. Eyedrops containing benzalkonium chloride (BAK; 0.01% wt/vol) and disodium edetate (EDTA; 0.01% wt/vol) showed maximum permeation, followed by Zymar, BAK (0.01% wt/vol), Gatilox, Gatiquin, and Gate (statistically significant P < .05 compared with control). In vitro titration of the formulations with 0.1N NaOH indicated the presence of a buffer in Zymar (pH 6) and Gate (pH 5.8), which may cause irritation and induce lacrimation, resulting in reduced ocular availability in vivo. Thus, formulation with BAK and EDTA, which is unbuffered, has a better likelihood of being absorbed in vivo. The BAK-EDTA formulation significantly (P < .05) increased the permeation of gatifloxacin through paired excised corneas of goat, sheep, and buffalo, compared with the control formulation. The goat cornea showed the greatest increase in permeation, followed by the sheep and buffalo corneas.

**KEYWORDS:** gatifloxacin, concentration, pH, preservative, cornea, permeation.

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#### INTRODUCTION

The efficacy of topical therapy with antibacterials (eg, sulfacetamide, chloramphenicol, polymixin B, gentamicin, fluoroquinolones) has been proven for infectious ocular conditions like conjunctivitis, keratitis, and endophthalmitis.<sup>1-3</sup> Fluoroquinolones, as a group, have shown excellent activity against the most frequently occurring gram-positive and gram-negative ocular pathogens.<sup>4-9</sup>

Fluoroquinolones have a unique mechanism of action: because of their strong affinity to the bacterial enzymes DNA gyrase and topoisomerase IV, they lethally interrupt bacterial replication. There appear to be fewer resistant bacterial strains against fluoroquinolones than against other antibacterials.<sup>10</sup> Earlier-generation fluoroquinolones such as ciprofloxacin and ofloxacin have been used widely to treat various pathogenic conditions. However, development of resistant strains against these fluoroquinolones has been reported.<sup>11,12</sup>

Gatifloxacin is a fourth-generation fluoroquinolone having a methoxy group in the C-8 position and a piperazine moiety at C-7. Gatifloxacin possesses an improved antibacterial spectrum, particularly against resistant staphylococcus and streptococcus pathogens, compared with older fluoroquinolones. Gatifloxacin has shown good aqueous solubility and better penetration than previous fluoroquinolones, so it has superior efficacy.<sup>13,14</sup> Studies have shown that topical gatifloxacin solution may be effective in prevention of infectious conditions such as conjunctivitis, keratitis, and postcataract endophthalmitis.<sup>15,16</sup> Owing to the benefits offered by gatifloxacin, various manufacturers have introduced gatifloxacin eyedrops (0.3% wt/vol). Fluoroquinolones have pHdependent solubility.<sup>17</sup> From the comfort and safety point of view, the eye can tolerate a limited range of pH. Preservatives/chelating agents are added to eyedrops primarily to maintain sterility during use.<sup>18</sup> The available eyedrops are formulated by different manufacturers, so they may vary in the type/concentration of additives and may have varying degrees of permeation through the cornea.

These studies examined the effect of formulation factors (eg, concentration, pH, presence of preservatives) on the corneal transport of gatifloxacin through isolated goat cornea. For the sake of comparison, the permeation characteristics of gatifloxacin from selected marketed formulations were evaluated using isolated goat cornea. Attempts were also made to evaluate the permeation characteristics of gatifloxacin from optimized and control formulations through paired corneas of goat, sheep, and buffalo.

# **MATERIALS AND METHODS**

Gatifloxacin sesquihydrate (purity 100% on an anhydrous basis) was obtained from Ranbaxy Laboratories Ltd (Gurgaon, India) as a gift. Benzalkonium chloride was obtained from Merck (Mumbai, India). The rest of the materials (analytical grade) were obtained from Central Drug House (New Delhi, India). Marketed gatifloxacin eyedrops—Gate (Ajanta Pharmaceuticals Ltd, Mumbai, India), Gatilox (Sun Pharmaceuticals Ind Ltd, Vapi, India), Gatiquin (Cipla Ltd, Verna, India), and Zymar (Allergan [India] Pvt Ltd, Pithampur, India)—were procured from a local market. Fresh eyeballs of goat, sheep, and buffalo were obtained from a local butcher shop (Ambedkar Nagar, New Delhi, India) within 1 hour of the animal's slaughtering. The method of dissection of cornea and the apparatus used in the permeation studies were the same as those described previously.<sup>19</sup>

# **PERMEATION EXPERIMENT**

Freshly excised goat cornea was fixed between clamped donor and receptor compartments of an all-glass modified Franz diffusion cell in such a way that its epithelial surface faced the donor compartment. The corneal area available for diffusion was  $0.67 \text{ cm}^2$ . The receptor compartment was filled with 10 mL of freshly prepared normal saline, and all air bubbles were expelled from the compartment. An aliquot (1 mL) of test solution was placed on the cornea, and the opening of the donor cell was sealed with a glass coverslip, while receptor fluid was kept at 37°C with constant stirring using a Teflon-coated magnetic stir bead. The permeation study was continued for 120 minutes; samples were withdrawn from the receptor and analyzed for gatifloxacin content by measuring absorbance at 291.5 nm in a spectrophotometer (1601 Shimadzu, Kyoto, Japan). Results were expressed as amount permeated and percent permeation or in vitro ocular availability. The permeation (%) or in vitro ocular availability was calculated as follows:

$$Permeation (\%) = \frac{Amount of drug permeated in receptor}{(Initial amount of drug in donor)} \times 100 (1)$$

At the end of the experiment, each cornea (freed from adhering sclera) was weighed, soaked in 1 mL of methanol, dried overnight at 90°C, and reweighed. From the difference of weights corneal hydration was calculated. The permeation characteristics of gatifloxacin from control and optimized formulations were also evaluated through freshly excised paired goat, sheep, and buffalo corneas. Statistical calculations were done by 1-way analysis of variance followed by Dunnett's test. A paired t test was used for studies with paired corneas. A P value less than .05 was considered significant.

# **PREPARATION OF TEST SOLUTIONS**

## Gatifloxacin Ophthalmic Solutions of Increasing Concentration of pH 7.2

The required amount of gatifloxacin was dissolved in a sufficient amount of distilled water, sodium chloride was added to make the final solution isotonic, the pH of the solution was adjusted to 7.2 using 0.1N NaOH and 0.1N HCl, and distilled water was added as needed to bring the final volume up to 100 mL, creating solutions of 0.1%, 0.2%, and 0.3% (wt/vol) concentration.

# Gatifloxacin Ophthalmic Solutions (0.3% wt/vol) of Different pH

Gatifloxacin (0.3 g) was dissolved in a sufficient amount of distilled water, sodium chloride was added to make the final solution isotonic, the pH of the solution was adjusted to 5.0, 6.0, 6.5, or 7.2 using 0.1N HCl and 0.1N NaOH, and distilled water was added as needed to bring the final volume up to 100 mL, creating solutions of different pH.

# Gatifloxacin Ophthalmic Solutions (0.3% wt/vol, pH 6.5) Containing Preservatives

Gatifloxacin (0.3 g) was dissolved in a sufficient amount of distilled water, sodium chloride was added to make the final solution isotonic, and the pH of the solution was adjusted to 6.5. To this solution benzalkonium chloride (BAK; 0.01% wt/vol), benzyl alcohol (BA; 0.5% vol/vol), thimerosal (THM; 0.005% wt/vol), phenyl mercuric acetate (PMA; 0.002% wt/vol), disodium edetate (EDTA; 0.01% wt/vol), or a combination of BAK (0.01% wt/vol) and EDTA (0.01% wt/vol) was added, and distilled water was added as needed to bring the final volume of each solution up to 100 mL.

### Surface Tension Measurement

The surface tension of the following ophthalmic solutions/ marketed eyedrop preparations was measured by using a stalagmometer: (1) gatifloxacin 0.3% wt/vol isotonic solution of pH 6.5; (2) gatifloxacin 0.3% wt/vol isotonic solution of pH 6.5 containing BAK (0.01% wt/vol); (3) gatifloxacin 0.3% wt/vol isotonic solution of pH 6.5 containing BAK (0.01% wt/vol) and EDTA (0.01% wt/vol); (4) Gate; (5) Gatilox; (6) Gatiquin; and (7) Zymar.

#### In Vitro Titration of Ophthalmic Formulations

One milliliter of each control formulation, ie control (without additive), optimized (containing BAK and EDTA) and marketed eyedrop, was titrated with 0.1N NaOH to a phenolphthalein endpoint.

### **RESULTS AND DISCUSSION**

The results of permeation studies through goat cornea (Table 1) showed that an increase in drug concentration in the drops resulted in an increase in amount permeated after 120 minutes, but the percent permeation and in vitro ocular availability decreased. The cornea has 3 layers: the epithelium, the stroma, and the endothelium. Only the amount of drug needed to saturate the epithelium would be able to partition through the stroma and endothelium to the receptor. Thus, an increase in concentration would have a negative effect on the in vitro ocular availability of the drug. Similar findings of reduced in vitro ocular availability with an increase in drug concentration have been reported for ibuprofen, flurbiprofen,<sup>20,21</sup> and moxifloxacin.<sup>22</sup> An increase in drug concentration did not affect the corneal hydration, however; it remained in the normal range of 75% to 80%.<sup>23</sup> An increase in moxifloxacin hydrochloride concentration from 0.1% to 0.3% (wt/vol) has been found to increase the amount permeated through excised goat cornea by 40% after 120 minutes,<sup>22</sup> whereas gatifloxacin showed a more than 100% increase in permeation under the identical conditions.

An increase in the pH of the ophthalmic solution from pH 5 to pH 7.2 increased both the amount of drug permeated and the in vitro ocular availability after 120 minutes (Table 1). Gatifloxacin has a pK<sub>a1</sub> of 5.94 for the carboxyl group and a pK<sub>a2</sub> of 9.21 for the piperazinyl group. The drug has a pI of 7.58. The octanol/water partition coefficient of gatifloxacin at pH 5.1 is 0.044, whereas at pH 7.0 the value increases to 0.145.<sup>24</sup> Thus, as the pH of the formulation is shifted toward neutrality (ie, the pH of tears), a larger fraction of the drug exists in an un-ionized state that is lipid-soluble. Having a higher lipid solubility at neutral pH would

promote gatifloxacin's permeation through the cornea. Corneal hydration was found to be 83.7% at pH 5, indicating that a lower formulation pH is damaging to the cornea. Corneal hydration remained in the normal range when the pH of the formulation was between 6.5 and 7.2. Transport of moxifloxacin, another fourth-generation fluoroquinolone, across excised goat cornea has also been found to be pH-dependent, with a maximum at pH 7.2,<sup>22</sup> but the pH-induced increase in permeation was much less marked compared with gatifloxacin's.

Evedrops are generally dispensed in multidose vials, each of which is intended to be used within a few days, once it is opened. Thus, to maintain the sterility of the preparation during use, a preservative is generally added to the formulation. Some common ophthalmic preservatives in their conventional concentrations were used in making the gatifloxacin 0.3% wt/vol formulation. The pH of the formulation was kept at 6.5, in consideration of the better solubility of the drug at lower pH that will impart physical stability (or prevent crystallization) at low ambient temperature. The effect of preservatives on the transcorneal permeation of the drug was evaluated using excised goat cornea. The results are shown in Table 2. Formulation with BAK (0.01% wt/vol) showed significantly (P < .05) higher permeation than did the control formulation with no preservative. Formulation with PMA (0.002% wt/vol), THM (0.005% wt/vol), and BA (0.5% vol/vol) produced lower permeation of drug than did the control formulation containing no preservative. EDTA (0.01% wt/vol), a known calcium chelator when used alone, did not increase the permeation of the drug through the cornea. The results are contrary to those observed with moxifloxacin transport through goat cornea.<sup>22</sup> BAK did not increase the permeation of moxifloxacin, whereas EDTA produced a significant increase in permeation. The physicochemical properties of the drug could affect corneal permeation. The combination of BAK and EDTA (each 0.01% wt/vol), however, increased the amount permeated and the percent permeation of gatifloxacin the most. Corneal hydration with the formulation containing BAK and

Table 1.	Effect of	Concentration	and pH	on I	Permeation	of	Gatifloxacin	Through	Excised	Goat	Cornea*
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Formulation Factor		Amount Permeated (mg) (120 minutes)	Permeation (%) (120 minutes)	Corneal Hydration (%)
Concentration (wt/vol) (gatifloxacin aqueous	0.1%	$0.016\pm0.00$	$1.60\pm0.04$	$75.9\pm0.65$
solution of different concentrations at pH 7.2)	0.2%	$0.028 \pm 0.001$ †	$1.40\pm0.06$	$77.9\pm0.48$
	0.3%	$0.035 \pm 0.002$ †	$1.17\pm0.06$	$75.9\pm0.65$
pH (gatifloxacin 0.3% wt/vol aqueous	5.0	$0.019\pm0.00$	$0.63\pm0.03$	$83.7\pm0.35$
solution at different pH)	6.0	$0.027 \pm 0.00 \ddagger$	$0.90\pm0.03$	$80.5\pm0.65$
	6.5	$0.029 \pm 0.00 \ddagger$	$0.97\pm0.03$	$79.1 \pm 0.41$
	7.2	$0.035 \pm 0.002$ ‡	$1.17\pm0.06$	$75.9\pm0.65$

\*Values are mean  $\pm$  SE of 3 corneas in each group. ANOVA indicates analysis of variance.

†Statistically significant (P < .05) compared with solution of 0.1% concentration as determined by 1-way ANOVA followed by Dunnett's test. ‡Statistically significant (P < .05) compared with solution of pH 5.0 as determined by 1-way ANOVA followed by Dunnett's test.

Table 2. Effect of Preservatives on Permeation of Gatifloxacin From 0.3% Aqueous Solution (pH 6.5) Through Excised Goat Cornea\*

	Amount Permeated (mg)	Permeation (%)	
Preservative	(120 minutes)	(120 minutes)	Corneal Hydration (%)
None (control)	$0.029\pm0.00$	$0.97\pm0.03$	$79.1 \pm 0.41$
PMA	$0.028 \pm 0.00$	$0.93 \pm 0.02$	$79.2 \pm 0.23$
THM	$0.025 \pm 0.00$	$0.83 \pm 0.01$	$80.6\pm0.63$
BAK	$0.071 \pm 0.003$ †	$2.37 \pm 0.11$	$77.9 \pm 1.35$
EDTA	$0.027 \pm 0.003$	$0.90\pm0.10$	$78.7\pm0.41$
BAK + EDTA	$0.083 \pm 0.002$ †	$2.76\pm0.08$	$78.3\pm0.81$
BA	$0.026 \pm 0.001$	$0.86\pm0.02$	$82.1 \pm 0.31$

\*Values are mean ± SE of 3 corneas in each group. BAK indicates benzalkonium chloride; EDTA, disodium edetate; BA, benzyl alcohol; THM, thimerosal; PMA, phenyl mercuric acetate.

 $\dagger$ Statistically significant ( $P \le .05$ ) compared with control, as determined by 1-way analysis of variance followed by Dunnett's test.

EDTA was in the normal range (78.3%), indicating no corneal damage, while the formulation containing BA showed corneal hydration of 82.1%, indicating slight corneal damage.

The addition of BAK (a cationic surfactant) reduced the surface tension of the gatifloxacin drop from 69.5 to 41.7 dynes/cm. The addition of EDTA to the formulation marginally reduced the surface tension (41.2 dynes/cm). The relative permeation characteristics of gatifloxacin from the optimized formulation containing BAK and EDTA, as well as from selected marketed preparations, are shown in Table 3 and Figure 1. The optimized formulation showed the highest permeation, followed by Zymar, Gatilox, Gatiquin, and Gate. Gatilox, Gatiquin, and Gate showed significantly (P < .05) higher permeation of drug compared with the control formulation, but their extent of permeation was much less compared with that of the optimized formulation or Zymar. The surface tension of the marketed formulations varied between 34.9 and 40.7 dynes/cm, indicating the presence of a surfactant. In fact, all the marketed evedrops contained BAK, which is known to increase the corneal permeation of the drug by disruption of the corneal epithelium.<sup>25</sup> It can also emulsify the corneal epithelium, leading to increased partitioning of the drug. The lower permeation of drug from Gatilox and Gate appears to be due to the lower pH of the formulation (pH 5.8), in which most of the drug would remain in the ionized state, while Gatiquin, which had a pH of 6.7, should have provided higher permeation, but a low concentration of BAK (0.005% wt/vol) might have contributed to its lower permeation. All 3 formulations showed corneal hydration between 80.6% and 81.2%, indicating slight corneal damage.

To ascertain the presence of buffer in marketed eyedrop formulations, 1 mL of each preparation was titrated with 0.1N NaOH to a phenolphthalein endpoint. Zymar had a pH around 6, and it consumed 3.35 mL of NaOH, while the control or formulation with BAK and EDTA consumed 0.05 mL of alkali. The result suggests the presence of a buffer in Zymar. Similarly, Gate contained a buffer and consumed 3.15 mL of NaOH. One might like to know the consequences of instilling a buffered drop in a living eye. Studies on ocular drops of pilocarpine, a weakly basic drug  $(pK_a = 7.1)$  used to treat glaucoma, might provide some answers. For maximum chemical stability, pilocarpine (hydrochloride or nitrate) drops are buffered to pH 4 to 5.<sup>26</sup> Ahmed and Patton<sup>27</sup> observed a 2-fold higher absorption of pilocarpine from an unbuffered drop of pH 4.5, instilled into rabbit eye, than from a drop buffered with 0.0667M phosphate buffer of pH 4.5. A subsequent study by Ahmed and Chaudhuri<sup>28</sup> also demonstrated higher ocular absorption of pilocarpine from an unbuffered drop of pH 4.0 than from drops of the same pH containing either 0.0667M phosphate

 Table 3. Relative Permeation Characteristics of Gatifloxacin From Control, Optimized (BAK + EDTA), and Selected Marketed

 Formulations Through Excised Goat Cornea\*

Formulation	pН	Amount Permeated (mg) (120 minutes)	Permeation (%) (120 minutes)	Corneal Hydration (%)	Surface Tension (dyne/cm)	0.1N NaOH Consumed (mL)
Control	6.5	$0.029\pm0.00$	$0.97\pm0.03$	$79.1 \pm 0.41$	69.5	0.05
Optimized	6.5	$0.083 \pm 0.002 \dagger$	$2.76\pm0.08$	$78.3 \pm 0.81$	41.2	0.05
Gate	5.8	$0.037\pm0.00\dagger$	$1.24\pm0.03$	$80.6\pm0.26$	35.5	3.15
Gatilox	5.8	$0.039\pm0.00\dagger$	$1.30\pm0.01$	$80.9\pm0.36$	36.5	0.15
Gatiquin	6.7	$0.039\pm0.00\dagger$	$1.30\pm0.03$	$81.2\pm0.77$	40.7	0.05
Zymar	6.0	$0.076\pm0.001\dagger$	$2.52\pm0.04$	$71.9 \pm 1.10$	34.9	3.35

\*Values are mean  $\pm$  SE of 3 corneas in each group.

 $\pm$  \$\text{Statistically significant (\$P < .05\$) compared with control as determined by 1-way analysis of variance followed by Dunnett's test.



Figure 1. Relative permeation characteristics of gatifloxacin from control, optimized (BAK + EDTA), and selected marketed formulations through excised goat cornea. \*Statistically significant (P < .05) compared with control as determined by 1-way analysis of variance followed by Dunnett's test. BAK indicates benzalkonium chloride; EDTA, disodium edetate.

or citrate or acetate buffer. Buffering a pilocarpine drop below the physiological pH of tears reduces ocular penetration both by suppressing the absorption of drug (as pilocarpine remains ionized at low pH) and by increasing reflex tear production and loss of drug. Zymar or Gate on instillation to the eye would have a similar fate. The presence of buffer would depress the postinstillation pH of lachrymal fluid, which would then be titrated back to physiological range by tears. Since the formulations have substantial alkali-consuming capacity, on instillation to the eye, they would cause irritation and induce lacrimation, resulting in an influx of tears and an increase in the drainage of the tears containing the drug away from the precorneal area or conjunctival sac. Thus, the formulation with BAK and EDTA, being unbuffered, is likely to be better absorbed than other formulations in vivo, but further studies are needed to confirm this. Zymar showed a corneal hydration of 71.9%, and the formulation appears to be slightly hypertonic.

The permeation characteristics of gatifloxacin from the optimized formulation containing BAK and EDTA and the control formulation were evaluated using paired corneas of goat, sheep, and buffalo. In these paired cornea tests, one



**Figure 2.** Relative permeation characteristics of gatifloxacin from control and optimized (BAK + EDTA) formulations through excised goat, sheep, and buffalo corneas (paired). \*Statistically significant (P < .05) compared with control as determined by paired *t* test. BAK indicates benzalkonium chloride; EDTA, disodium edetate.

cornea from a single animal was treated with the optimized formulation while the same animal's other cornea was treated with the control formulation containing no additive. This procedure was adopted to minimize biological variation. The results (Table 4, Figure 2) showed that, for all the mammalian corneas, the formulation containing BAK and EDTA significantly (P < .05) increased the permeation of gatifloxacin compared with the control formulation. The maximum increase in drug permeation was observed with goat cornea, which had the minimum thickness, and the minimum increase was observed with buffalo cornea, which had the maximum thickness. Corneal hydration levels with sheep and buffalo corneas were in the normal range (ie, 75%-80%), while with goat cornea the hydration was 80.7%, indicating slight corneal damage. Since the corneal hydration level was below 83%, the damage appears to be reversible. Thus, the optimized formulation could be considered safe and nondamaging to the eye. The combination of BAK and EDTA has also been reported to increase the permeation of moxifloxacin through excised goat, sheep, and buffalo corneas.<sup>22</sup>

**Table 4.** Relative Permeation Characteristics of Gatifloxacin From Control and Optimized Formulations Through Excised Goat, Sheep, and Buffalo Corneas (Paired)\*

		Control Fo	ormulation (No Add	ditive)	Optimized Formulation (BAK + EDTA)			
Animal	Thickness of the Cornea (mm)	Amount Permeated (mg) (120 minutes)	Permeation (%) (120 minutes)	Corneal Hydration (%)	Amount Permeated (mg) (120 minutes)	Permeation (%) (120 minutes)	Corneal Hydration (%)	
Goat Sheep Buffalo	$\begin{array}{l} 0.66 \pm 0.005 \\ 0.80 \pm 0.011 \\ 1.12 \pm 0.011 \end{array}$	$\begin{array}{l} 0.022 \pm 0.001 \\ 0.021 \pm 0.001 \\ 0.020 \pm 0.002 \end{array}$	$\begin{array}{c} 0.73 \pm 0.06 \\ 0.70 \pm 0.05 \\ 0.66 \pm 0.06 \end{array}$	$\begin{array}{c} 80.2 \pm 0.40 \\ 77.3 \pm 0.79 \\ 78.1 \pm 0.31 \end{array}$	$\begin{array}{l} 0.076 \pm 0.088 \dagger \\ 0.060 \pm 0.003 \dagger \\ 0.052 \pm 0.004 \dagger \end{array}$	$\begin{array}{c} 2.53 \pm 0.09 \\ 2.0 \pm 0.10 \\ 1.73 \pm 0.15 \end{array}$	$\begin{array}{l} 80.7 \pm 0.06 \\ 77.0 \pm 1.87 \\ 79.1 \pm 0.23 \end{array}$	

\*Values are mean  $\pm$  SE of 3 corneas in each group.

 $\dagger$ Statistically significant (P < .05) compared with control as determined by paired t test.

#### CONCLUSIONS

On the basis of these studies, it can be concluded that gatifloxacin 0.3% (wt/vol) ophthalmic solution (pH 6.5) containing BAK (0.01% wt/vol) and EDTA (0.01% wt/vol) provides the maximum in vitro ocular availability through goat cornea. The formulation also increased the permeation of gatifloxacin through excised sheep and buffalo corneas. Among the marketed formulations, Zymar showed the highest permeation, but the low pH of and presence of buffer in Zymar could increase its ocular irritation potential, which could result in lacrimation and reduced ocular availability in vivo.

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